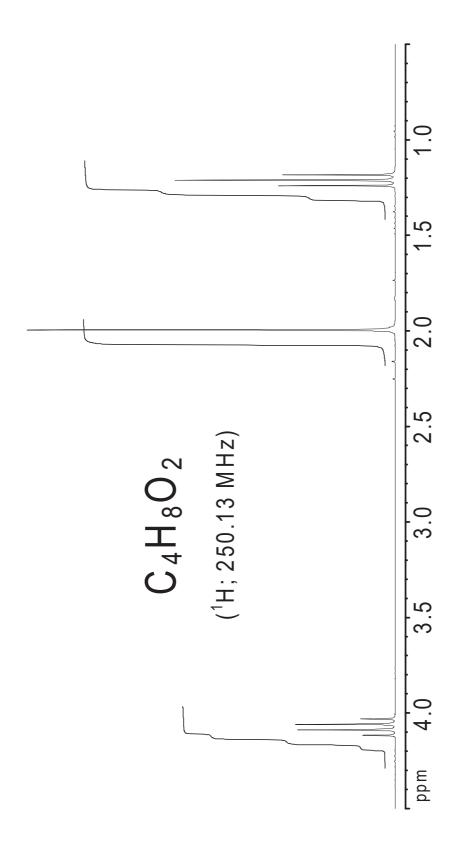
Problem 1

Translated from Art van der Est)

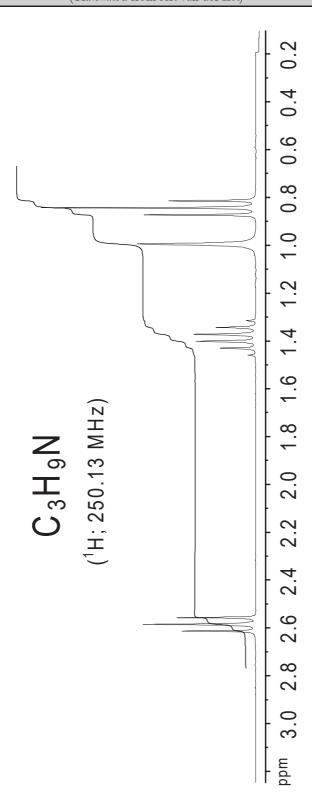
Using the chemical shift rules for CH-, CH₂- und CH₃-Groups draw the stick spectrum for propionic acid chloride. There is no shift rule for acid chloride groups. Search for an acceptable alternative.



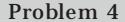


Exercises, problem 2: Determine the structure. Note that the integrals have been shifted a few millimeteres to the left for clarity.

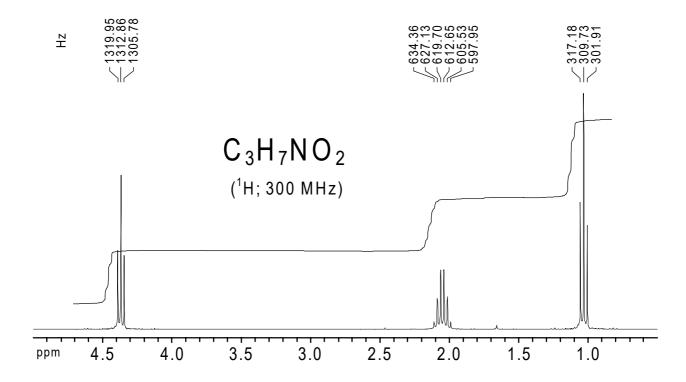




Exercises, problem 3: Determine the structure! Explain the splitting patterns! How large do the coupling constants have to be? Estimate the chemical shifts using the appropriate shift rules!

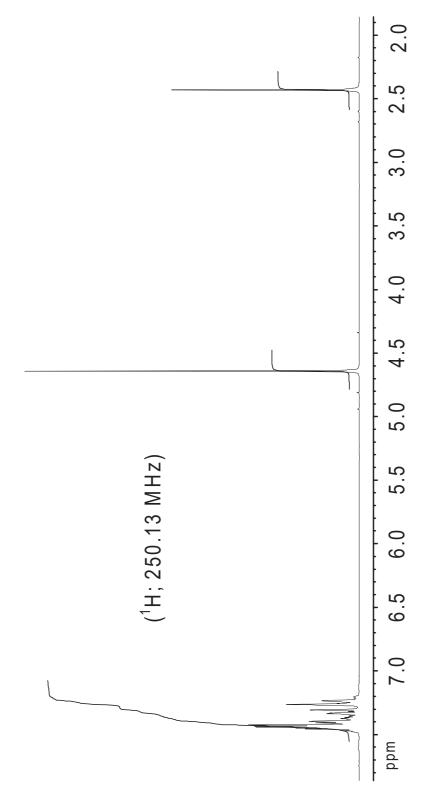


(Translated from Art van der Est)

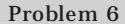


Exercises, problem 4: Determine the structure. Why is a clean sextet observed at 2 ppm? What is the actual multiplicity? The data are from a collection of the Pacific Lutheran University, Washington.

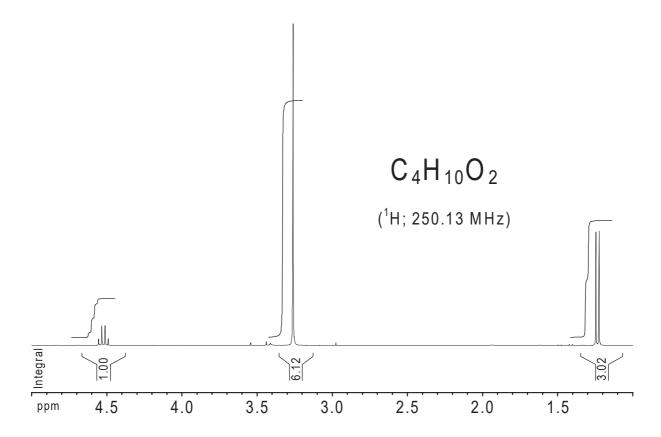




Exercises, problem 5: During the photo-chlorination of toluene, a mixture of products is obtained. Using the proton spectrum, determine the two components! In what mole ratio are the present?

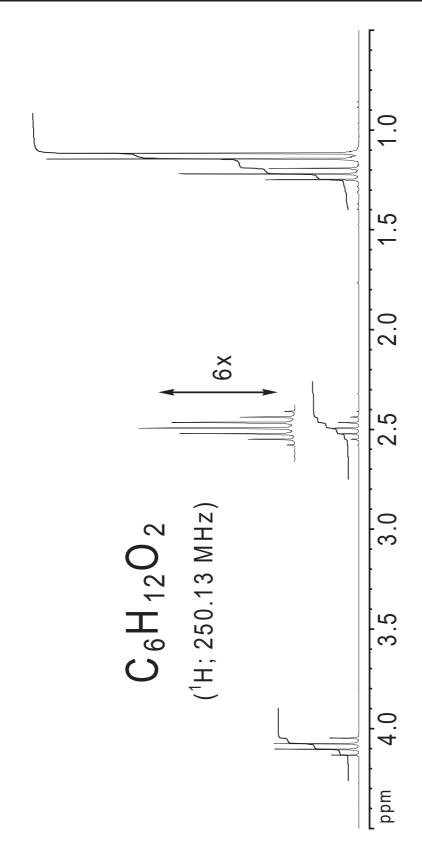


(Translated from Art van der Est)

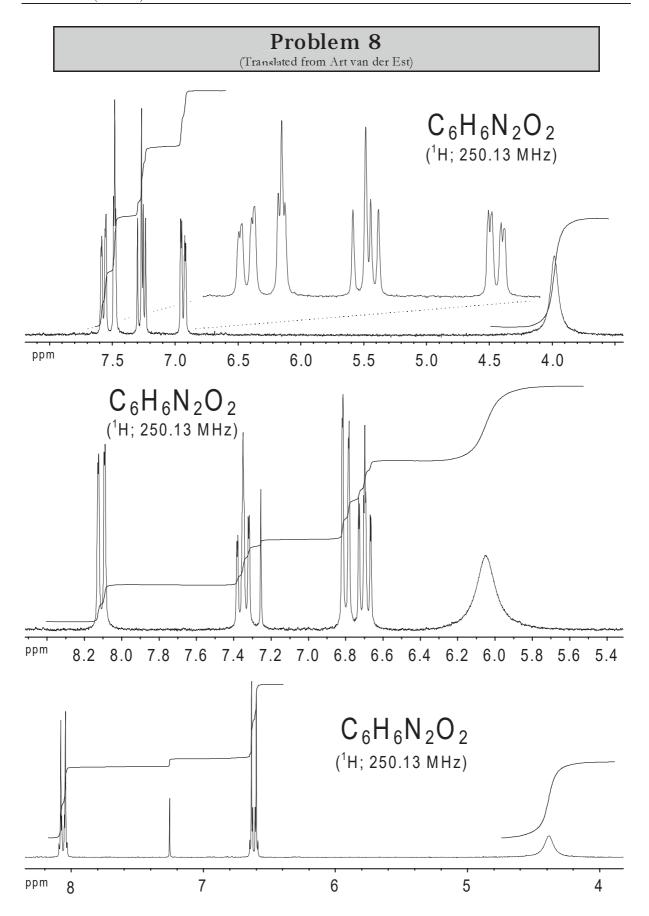


Exercises, problem 6: Determine the structure. Note that the integrals have been shifted a few millimeteres to the left for clarity.

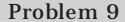




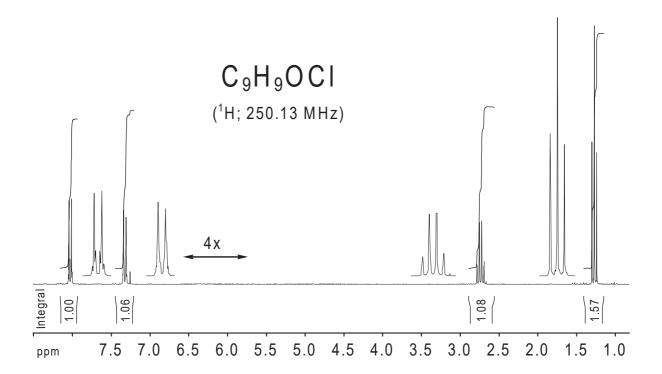
Exercises, problem 7: Determine the structure. Check the chemical shifts using shift rules!



Exercises, problem 8: Distinguish the three isomers of a disubstituted benzene derivative! Also interpret the chemical shift of the broad Signal! Which solvent was used?



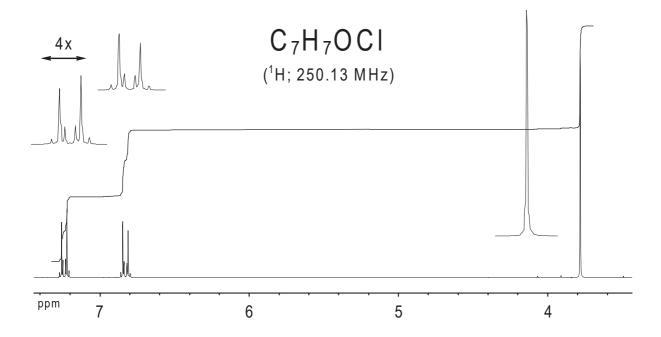
(Translated from Art van der Est)



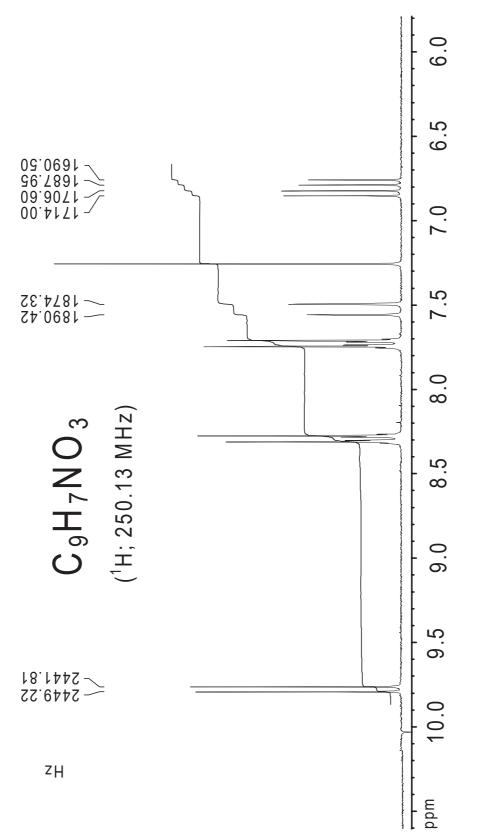
Exercises, problem 9: Determine the structural formula using the NMR-spectrum and the molecular formula!



(Translated from Art van der Est)

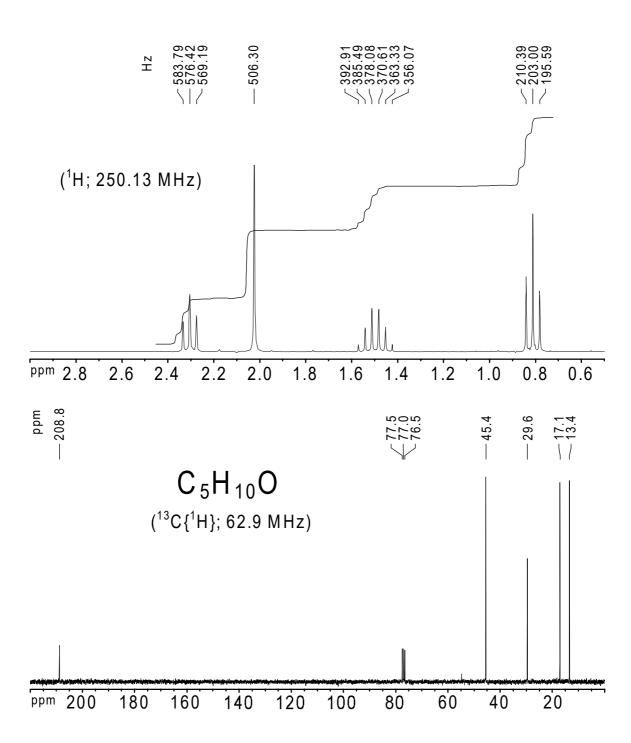




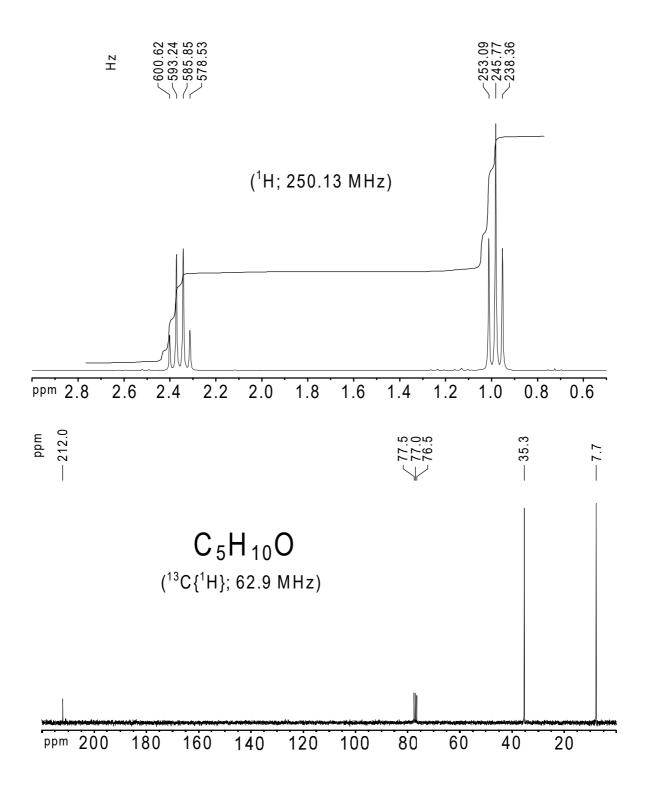


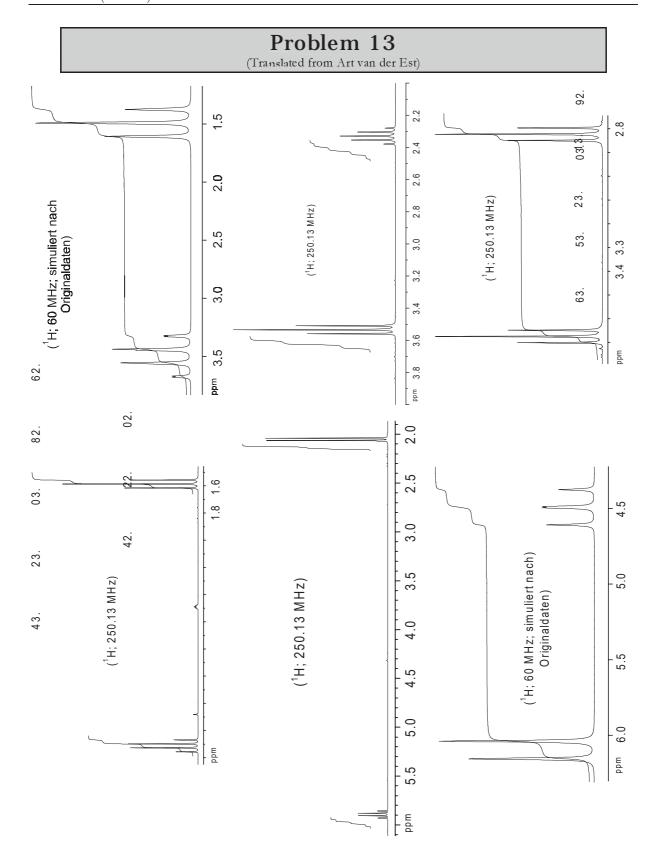
Exercises, problem 11: Determine the structure and the conformation!





Exercises, problem 12 (page 1): Determine the structure of the two structural isomers (see additional spectra on second page)!

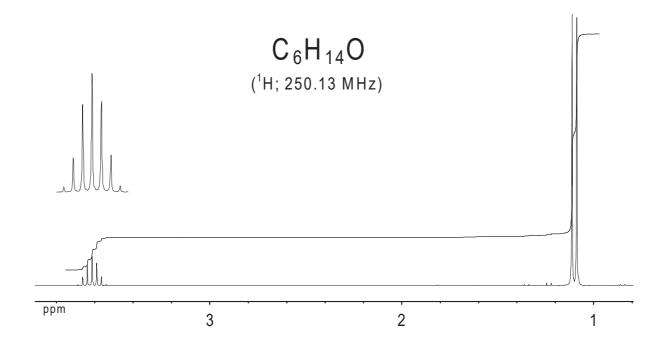




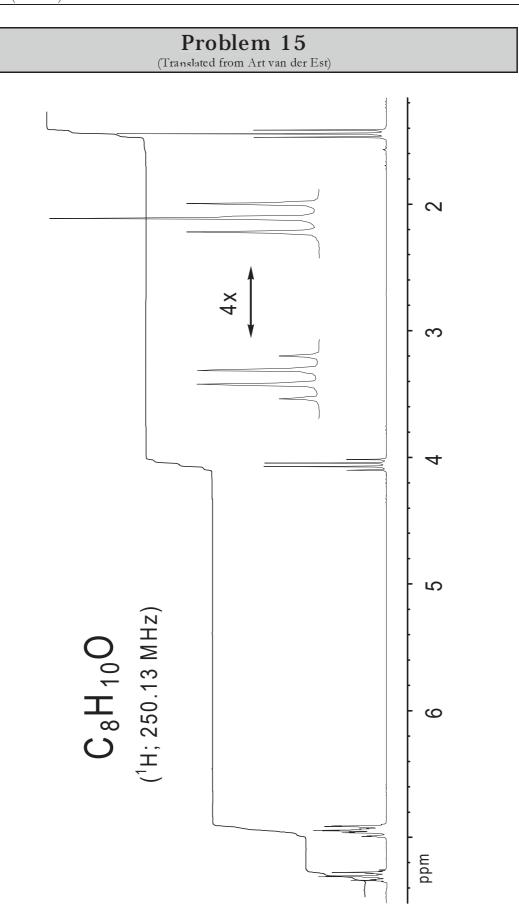
Exercises, problem 13: The 6 spectra correlate with the following structural formulae: CH₃-CH₂-OH, Cl₂CH-CH₂-CHCl₂, (CH₃)₂-CH-CN, CHCl₂-CHCl-CHCl₂, CHCl₂-CHO, N≡C-CH₂-Cl, CHCl₂-CHCl₂, CH₃-CH₂-Cl, CH₂Br-CH₂-CH₂Br, CH₃-CH₂Br, CH₃-CHCl₂ und (CH₃)₂-CHCl. Assign them correctly and explain your choice!



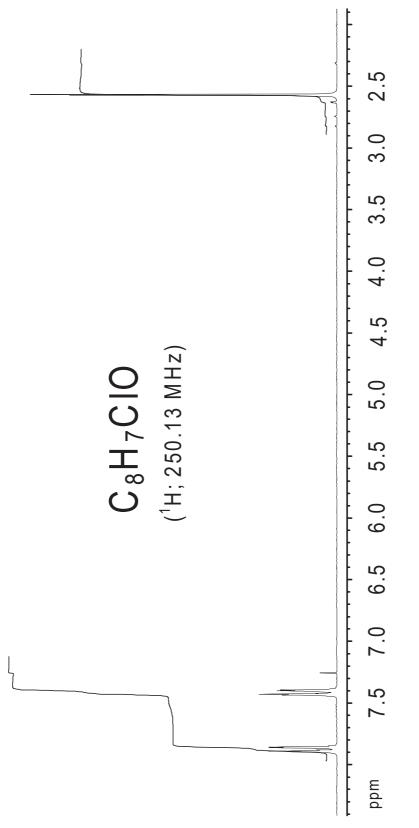
(Translated from Art van der Est)



Exercises, problem 14: Determine the structure. Hint: Work out the multiplicity of the multiplets at about 3.6 ppm! How can this be explained?



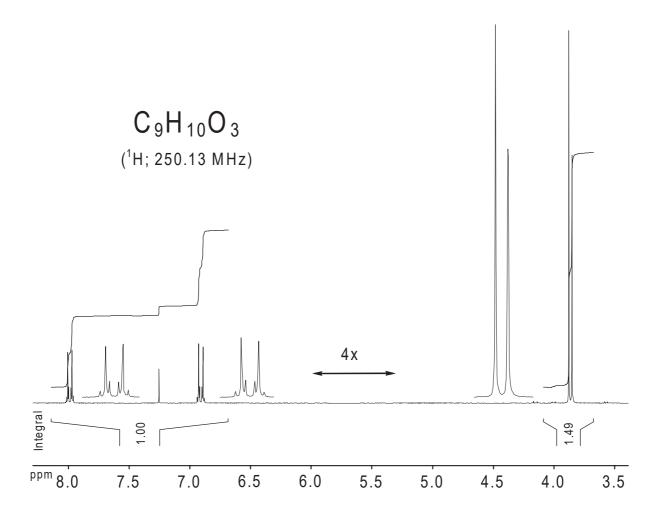




Exercises, problem 16: Determine the structure! Can you distinguish reliably between the two alternative structures?



(Translated from Art van der Est)



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